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Based on Sanskrit mathematical aphorisms found in the Atharvaveda.

This book had its nucleus in some lectures given by one of us (J. O'M. B.) in a course on electrochemistry to students of energy conversion at the University of Pennsylvania. It was there that he met a number of people trained in chemistry, physics, biology, metallurgy, and materials science, all of whom wanted to know something about electrochemistry. The concept of writing a book about electrochemistry which could be understood by people with very varied backgrounds was thereby engendered. The lectures were recorded and written up by Dr. Klaus Muller as a 293-page manuscript. At a later stage, A. K. N. R. joined the effort; it was decided to make a fresh start and to write a much more comprehensive text. Of methods for direct energy conversion, the electrochemical one is the most advanced and seems the most likely to become of considerable importance. Thus, conversion to electrochemically powered transportation systems appears to be an important step by means of which the difficulties of air pollution and the effects of an increasing concentration in the atmosphere of carbon dioxide may be met. Corrosion is recognized as having an electrochemical basis. The synthesis of nylon now contains an important electrochemical stage. Some central biological mechanisms have been shown to take place by means of electrochemical reactions. A number of American organizations have recently recommended greatly increased activity in training and research in electrochemistry at universities in the United States.

Electrochemistry is a collection of papers presented at the First Australian Conference on Electrochemistry, held in Sydney on February 13-15 and in Hobart on February 18-20, 1963, jointly sponsored by The Royal Australian Chemical Institute, The University of New South Wales, and The University of Tasmania. This conference highlights the numerous advances in the field of electrochemistry. This book is organized into 12 parts encompassing 70 chapters. The first parts deal with the solid-state reactions and processes in electrochemistry; the thermodynamic aspects of electrolytes; and the role of electrochemistry in corrosion control. The succeeding parts explore the concepts of equilibrium and non-equilibrium theory of double layers, as well as the various electroanalytical methods used in electrochemistry, including polarography, potentiometry, and coulometry. Other parts consider the areas of application of electrochemistry, such as in electroplating, anodizing, fuel cell, electrowinning, and electrorefining. The remaining chapters are devoted to non-aqueous electrolytes, molten salts, and electrode and electrochemical processes. Electrochemists and physicists will find this book invaluable.

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