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Answer the following questions in
the space provided. 1. The

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Following are statements about the dissolving process. Explain each one at the molecular level.

a. Increasing the pressure of a solute gas above a liquid solution increases the solubility of the gas in the liquid.

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A homogeneous mixture that contains particles in a dispersed phase that do not settle out is a solution. The solution process with water as the solvent. hydration. Liquid solutes and

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solvents that are not soluble in each other are. immiscible. A solute molecule that is surrounded by solvent molecules is. solvated.

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homogeneous mixture of two or more substances, consisting of ions or molecules. solute. gas or solid dissolved in a liquid or solution component in small amount. solvent. liquid that

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dissolves a gas or solid or the solution component in greater amount.

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SECTION 1 SHORT ANSWER

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Answer the following questions in the space provided. 1. Match the type of mixture on the left to its representative particle diameter on the right. c solutions (a) larger than 1000 nm a suspensions (b) 1 nm to 1000 nm b colloids (c) smaller than 1 nm 2. Identify the

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solvent in each of the following examples:

~~mc06se cFMs r i vi~~

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56 2) = 4, = 10, = 16, = 22, = 28

3) = 4, = 8, = , = 64, = 4) an =

2n - 3 5) = \square 4n 6) = 5, = 3, = 1,

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8) 9) 15 10) $d = -5$; $a = 4$, $a_n = -1$, $S_n = -6$, $S_{10} = -11$ 11) $S_n = 94 - 10n$; $S_{10} = 14$ 12) 220 13) $S_n = 73$, $d = -5$, $a_n = -5$ 14) 34980 15) 25

~~Chapter 12 Review Solutions, $a_n = 10$, $S_n = 16$, $S_{10} = 22$, $S_{10} = 28 + 1) = 2$.~~

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CHAPTER 12. REVIEW "I" - Chemistry

Solutions . SHORT ANSWER

Answer the following questions in the space provided. 1. Describe the errors made by the following students in making molar solutions. 3. James needs a 0.600 M solution of KCl. He measures

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out 0.600 g of KCl and adds 1 L of water to the solid. James made several errors.

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SECTION 3 SHORT ANSWER

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_____ water to the solid. _____
_____ b. Mary needs a 0.02
M solution of NaNO_3 . She
calculates that she needs 2.00

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colloids.

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INSTRUCTIONS: Write your answers on lined or blank sheets of paper. You do not need to write

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the question, ONLY WRITE THE PROBLEM NUMBER and ANSWERS/SOLUTIONS. □ For problems that involve calculations, you must show your work to get full credit.

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mixtures of two or more
substances in a single phase.

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Soluble describes a substance as capable of being dissolved.

Solvent is the dissolving medium in a solution.

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12-Solutions Section 1- Types of Mixtures Solutions are homogeneous mixtures of two or more substances in a single phase. Soluble describes a substance as capable of being dissolved. Solvent is the dissolving medium in a solution.

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Solute is the substance that is dissolved in a solution.

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