

Chapter 4 Atomic Structure Test A Answers

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Matching. A. Bohr B. Democritus C. Rutherford D. Dalton E. Thomson F. Schrodinger.

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_____ 1. Greek thinker; called nature 's basic particle an atom, based on the Greek word " atomos " which means " indivisible ". Did not have evidence that atoms existed.

CHAPTER 4 TEST: Atoms, Atomic Theory and Atomic Structure

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Questions : 1. The number of electrons in Na^+ (atomic number = 11) are (a) 11 (b) 10 (c) 12 (d) 13 2. There are four elements P, Q, R and S having atomic numbers of 4, 18, 10 and 16 respectively. The element which can exhibit covalency as well as electrovalency will be: (a)...

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The four atomic species can be represented as follows. Out of these, the two species which can be termed isobars are: (i) $^{201}\text{X}_{60}$ (ii) $^{200}\text{X}_{61}$ (iii) $^{200}\text{X}_{58}$ (iv) $^{203}\text{X}_{60}$

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Atomic Structure (Chapter Test A) STUDY. Flashcards. Learn. Write. Spell. Test. PLAY. Match. Gravity. Created by. emily_reidt. Terms in this set (21) the total number of protons and neutrons in the nucleus of an atom. mass number. the weighted average mass of the atoms in a naturally occurring sample of an element.

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mixtures of whole number ratio compounds. 4) Chemical reactions occur when atoms are separated, joined or rearranged.

Chemistry Chapter 4 Atomic Structure Test

Weighted Atomic mass is $26.4 + 8.9 = \underline{\hspace{2cm}}$ amu. Title: Notes Chapter 4: The Structure of the Atom Author: Administrator Last modified by: Kelli Pearce Created Date: 9/15/2015 10:36:00 PM Company: MISD Other titles: Notes Chapter 4: The Structure of the Atom ...

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