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Education. Laidler received his early education at Liverpool College. He received his BA (1934) and MA (1938) degrees from Trinity

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College, Oxford University. His MA was in the area of chemical kinetics under Cyril Norman Hinshelwood. He completed his PhD in 1940 from Princeton University, with a thesis entitled: The Kinetics of Reactions in Condensed and Heterogeneous Systems, under Henry Eyring.

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(i) Consider the rate law $r = k[H_2][I_2]$. If we substitute units into the equation, we obtain $(\text{mol dm}^{-3} \text{ s}^{-1}) = [k] (\text{mol dm}^{-3}) (\text{mol dm}^{-3})$ where the notation $[k]$ means 'the units of k '. We can rearrange this expression to find the units of the rate constant, k . $[k] = (\text{mol dm}^{-3} \text{ s}^{-1}) (\text{mol dm}^{-3}) (\text{mol dm}^{-3}) = \text{mol}^{-1} \text{ dm}^3 \text{ s}^{-1}$

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