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Introduction to Conversion

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Residence Time Distribution

for Ideal Reactors

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Reactor Sequence: CSTR and

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Octave Levenspiel was a
professor of the field
Chemical engineering at
Oregon State University. In
this vast and evergreen

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1 Chemical reactions 1.1
Rate of reaction and
dependence on temperature We
will once again look at the
formation of ammonia (NH_3)
from nitrogen and hydrogen
(see section Chemical
equilibrium of the

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thermodynamics chapter).

This reaction follows the
equation: $N_2 + 3H_2 \rightarrow 2NH_3$

(1) $H_f^\circ = 92 \text{ kJ mol}^{-1}$ $S_f^\circ = 192$
J mol⁻¹ K⁻¹ To find the Gibbs free
energy of ...

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By sizing a chemical reactor we mean we're either determining the reactor volume to achieve a given conversion or determine the conversion that can be achieved in a given reactor type and size. Here we will assume that we will be given $-r_A = f(X)$ and F_{A0} . In chapter 3 we show how to find $-r_A = f(X)$. Given $-r$

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As a function of conversion, $-r_A = f(X)$, one can size any type of reactor.

Elements of Chemical Reaction Engineering

Octave Levenspiel was a professor of chemical engineering at Oregon State University. His principal interest was chemical reaction engineering, and he was the author of a major textbook Chemical Reaction Engineering as well as numerous research publications.

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