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An electron is contained in a three-dimensional rectangular box given by  $0 \leq x \leq a$ ,  $0 \leq y \leq b$ , and  $0 \leq z \leq c$ . The solutions of the Schrödinger equation are specified by the quantum numbers  $n_x$ ,  $n_y$ , and  $n_z$ .

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Recall that the normalized wave function is...with energy...where  $n_x$ ,  $n_y$ , and  $n_z$  are positive integers.

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$l(l+1) \rightarrow m^2 - 1/4$ , which is to say,  $l^2 + l + 1/4 = m^2$ , or  $(l + 1/2)^2 = m^2$ , or  $l = |m| - 1/2$ . [Our present equation depends only on  $m^2$ , and hence is the same for either sign, but the solution to Problem 4.39 assumed  $l + 1/2 \geq 0$  (else  $u$  is not normalizable), so we need  $|m|$  here.] Quoting 4.39:  $E = (j.$

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